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Teaching

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Chemistry
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Chapter 9
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Chapter 9

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Slavery EduDay
2020: Stylus
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Choice Tests

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Don't Cheat

Starting Your Own
Law Firm? Do
These 3 Things
First...Gaining the
Courage to Start
Your Own Firm Full
Life Cycle Use
Case: Bill Aulet (Pt

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1) How to Ace Multiple Choice Tests
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How to write English assignment | Class 6 | Lesson -1 | HW A1 - B4
What Is Justice?: Crash Course Philosophy #40
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- MATERIALS AND

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THINGS PART - 01

- SORTING OF
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WORKSHEET Ionic

Bonds Use with

Chapter 7, Section

7.2 1. How many

valence electrons

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does a neutral magnesium (Mg) atom have? 2. What is the charge on a magnesium ion?

What does magnesium have to do to form such an ion, and why does it tend to do so? 3.

How many valence electrons does a single neutral chlorine (Cl) atom

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have? 4. Transparency

Worksheets

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Chapter 9

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Chemistry: Matter
and Change Chapter

5 . uest 11 Name:

Date: Hour:

Information: Energy

Levels and

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Sublevels ...

Considering your answer to question 5, how many

electrons can fit in the entire 5th energy level? can (A VeI is a ha AC5a) 5 0 3 7.

Write the notation for an electron spinning clockwise in a p sublevel ...

Grosse Pointe

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WORKSHEET The
Activity Series Use
with Chapter 9,
Section 9.2 1. For
each of the
following pairs of
elements, underline
the one that would
replace the other

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element in a
compound. a.
calcium, tin e. iron,
copper b. bromine,
fluorine f. iodine,
chlorine c.
aluminum,
potassium g. silver,
lead d. zinc, sodium
2.

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WORKSHEET The

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Chemistry: Matter and Change 0

Chapter 8 . 1.7, the
CF bond is polar
covalent. (b)

$$|\text{ENCl} - \text{P}| = |\text{ENCl} -$$

$$\text{ENP}| = 3.0 - 2.1$$

$$|\text{ENCl} - \text{P}| = 0.9.$$

Since $0.5 < \text{EN} . \text{A} .$

< 1.7 , the PCl bond
is polar covalent.

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(c) !ENCI"Li

Section 8.5

Electronegativity
and Polarity 265

Compare and

contrast polar . that
is in excess.

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WORKSHEET

Atomic Orbitals 1.

What is the shape
of an s orbital?

Class 16 Use with
Chapter 5, Section

5.2 2. What is the
relationship

between the size of

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an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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does matter matter,
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Exponents work for
Chapter 9
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Redox reactions,
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master 41,

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key chapter 19,

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3 matterproperties

and changes, The ...

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