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- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC-COOH}\cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of NaOH (aq) . 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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5. a ratio of electrons : aluminium ions 3 : 1 amount Al 13 1 1.296 10 C 96500Cmol 3 «» 4.48 107 «mol» mass Al « 4.48 107 mol 26.98 g mol 1» 1.21 109 «g» Award [3] for correct final answer. 3 5.

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Deduct 1 mark in the paper for an error of 2 or more digits unless directed otherwise in the markscheme. e.g. if the answer is 1.63: 2 reject 1.6 accept 1.63 accept 1.631 accept 1.6314 reject Indicate the mark deduction by writing -1(SD)at the first point it occurs and SD on the cover page.

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